Quantitative Predictability of Carcinogenicity of the Covalent Binding Index of Chemicals to DNA: Comparison of the *In Vivo* and *In Vitro* Assays

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The capability of covalent binding to DNA to predict the initiating potential of chemical carcinogens was compared for the assays performed in vivo (rodent liver DNA) and in vitro (purified DNA incubated in the presence of mouse and rat liver microsomes). A quantitative correlation between DNA adducts and carcinogenic potency was investigated. The in vivo assay appeared slightly, but not significantly, more predictive than the in vitro assay. Also predictivity was slightly higher both in vivo and in vitro when we referred to liver carcinogenicity instead of overall carcinogenicity. The predictive ability found for DNA covalent binding (both in vivo and in vitro) was similar to that of many short-term tests (such as mutagenicity, DNA damage/repair, SCEs, and cell transformation tests).

The covalent DNA binding, measured after incubation with DNA in vitro in the presence of liver microsomes, could therfore be a reasonable short-term test offering greater rapidity of execution and requiring the sacrifice of fewer animals than the corresponding in vivo test.

Introduction

From our knowledge of the carcinogenicity process, genotoxicity seems to be especially related to initiation. Possibly, other irreversible genomic alterations occurring during progression can also be related to genotoxic effects. Considering that genotoxicity is related only to one or a few steps of the carcinogenicity process, it is not surprising that the correlation between carcinogenicity and genotoxicity is far from perfect.

In addition, when a genotoxicity test is performed, very often important discrepancies are found between the metabolism in the test system and real metabolism in the target cells for the initiation process. Even the end points of genotoxicity tests are not necessarily coincident with events relevant to the initiation process.

between the carcinogenic process and genotoxic events is reflected by the results of several studies investigating correlations between carcinogenicity and short-term tests (1-5).

In the papers published around 1975 and 1976, that

This relatively weak and partial correlation

examined very strong genotoxic agents and very clear-cut nongenotoxic compounds, a qualitative correlation around 90% was observed between shortterm test results and carcinogenicity (1-3). In subsequent papers the average level of correlation was around 70 to 80%. A global picture of the situation is offered from a study performed in 1984 by the International Commission for the Protection Against Environmental Mutagens and Carcinogens (ICPEMC) comparing carcinogenicity and genotoxicity in several short-term tests. In Table 2 of that study, the mean of the accuracy values (correct matchings) for 31 different tests was around 78% (4). In a more recent assessment by Tennant et al. (5), the level of qualitative correlation (accuracy) was only about 60% for four different in vitro short-terms tests. In the data base used for this assessment, the number of compounds suspected of being nongenotoxic carcinogens and/or promoting agents was particularly significant. Other authors (6-8) have expressed the opinion that when

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the fraction of nongenotoxic chemicals goes up in a data base the correlation between carcinogenicity and short-term tests goes down.

Studies of correlation between carcinogenic potency and quantitative degrees of response for positive short-term test data have given similar results (9-14). Again, a correlation is present, but relatively weak. For a summary of the quantitative correlation studies, see Table 2 in our previous paper (15). In that paper we also discussed the mathematical relationship between qualitative and quantitative correlation studies (15). The fact that a quantitative correlation is always present, even if weak, suggests that the additive role of promotion can attenuate, but not completely eliminate, the quantitative relationship existing between initiating potency and carcinogenic potency, provided that the bulk of the data base analyzed is not made up of pure, nongenotoxic chemicals.

As suggested above, differences between metabolism in the activation system of the short-term test and metabolism in the target cells for the initiation process could be a second major cause of the weak predictability of a given short-term test.

In the present study we tried to assess the relevance of metabolic activation in vivo in liver cells, versus an in vitro system of microsomes, again obtained from liver cells. The amount of DNA adducts was the common end point for both the in vivo and the in vitro assays considered, so the reason for a possible different degree of predictivity of the two tests had to be essentially related to the different metabolic activation. In addition, two other important factors could decrease the relationship between the in vivo and in vitro assays: first, DNA repair in the whole liver system may modulate the final DNA binding measured; second, there may be a difference in reactivity between chromosomal DNA and the purified DNA employed in the in vitro assay.

We had reason to suspect that a difference could indeed exist, considering the results obtained in several studies which we report below.

For instance, we compared the results obtained by Mirsalis and Butterworth (16), who examined the autoradiographic repair induced by dimethylnitrosamine (DMN) in rat liver cells after treatment in vivo and also the results obtained by Williams and Laspia (17), who examined the autoradiographic repair induced by DMN in primary cultures of hepatocytes where DMN was already clearly active in vivo at 10⁻⁵ mole/kg, but active in vitro only at 10⁻² M concentration.

In a paper by Kerklaan et al. (18), the induction of repairable DNA damage in *E. coli* cells evaluated in a host-mediated assay and in an *in vitro* assay were compared. The effects of DMN and 1,2-dimethyl-hydrazine (1,2-DMH) in *E. coli* cells recovered from the liver of injected mice were already clearly detectable at the dosage of 50 µmole/kg. In the *in vitro* assay (with the presence of mouse liver microsomes),

DMN showed an extremely weak (not dose-dependent) response in a range of concentrations between 10^{-2} and 10^{-1} M; 1,2-DMH was completely inactive in the *in vitro* assay up to a concentration of 50 mM.

In our comparison between adduct formation in vivo and in vitro, the data available were mostly positive. Because the qualitative approach requires a balanced presence of positive and negative results, we were forced to choose a quantitative approach in our correlation study. We already used this methodolgy in several investigations (9-13). On the other hand, as a reference point, we had a paper by Lutz presenting a detailed quantitative study of the correlation between carcinogenicity and DNA adduct formation in vivo (14).

Methods

The criteria adopted for the selection and computation of the data and for their correlation studies are discussed in the following section.

Carcinogenic Potency

The carcinogenic potency data were mainly obtained from the data base of Gold et al. (19). We can expect a homogenous computation of potencies for all the chemicals listed in it. In this data base the carcinogenic potency is evaluated as TD_{50} and is defined as "that chronic dose rate (in mg/kg body weight/day), which would halve the actuarially adjusted percentage of tumor-free animals at the end of a standard experiment time—the standard lifespan for the species" (20). For our computations, the TD_{50} values were normalized in terms of μ mole administered per kilogram body weight. For a minor group of chemicals, the above-mentioned data base was integrated with our own data (21). Our computations were brought in line with those of Gold et al. (19).

For each of the papers listed in the data base, we used the lowest reported TD_{50} value (maximal oncogenic potency) where carcinogenicity was considered statistically significant (p < 0.05). When more than one paper was reported in the data base for the same chemical, one TD_{50} value was selected from each of the papers. We then calculated the average of log TD_{50} .

Values of TD_{50} for liver tumors were also considered for correlation studies. The liver was the organ where DNA adducts were measured.

Data were collected only from experiments on mice and rats after oral or parenteral administration. Only positive data were considered.

For our purposes, all the chemicals studied for the correlation between in vitro and in vivo DNA covalent binding and oncogenic potency had to be genotoxic. We defined as genotoxic the chemicals reported as positive in at least one-third of the genotoxicity tests considered in the data base of PaPajda and Rosenkranz (22). This partial arbitrary cut-off point was established in order to exclude a

study that could be suspected as promoting carcinogens more so than initiating carcinogens. One such example could be diethylstilbestrol. We are aware that this cut-off is an arbitrary one. However, it is well known that genotoxic chemicals are not positive in 100% of the tests. For instance, Ashby et al. estimated that nongenotoxic chemicals could be positive in about 20% of the tests, and very genotoxic chemicals could be positive in about 80% of the tests (23). Less potent genotoxic agents could give a response between 80 and 20%. The level selected was one considered capable of excluding a significant fraction of nongenotoxic chemicals without being too strict about the genotoxic efficiency required. We were not able to find a more objective threshold. On the other hand, too much time and detailed data was required to analyze the quality of each manuscript and, if necessary, to disagree with the results of the authors about the initiating potential of the chemicals considered.

In Vivo DNA Covalent Binding (In Vivo-CBI)

The data concerning the *in vivo* production of DNA adducts were obtained mostly from papers by Lutz (14, 24); some other data were obtained from the papers used for computing the *in vitro* DNA covalent binding index. According to Lutz (24), the *in vivo*-CBI was defined as:

micromole chemical bound per mole nucleotides
millimole chemical administered per kilogram animal

The *in vivo*-CBI was computed for experiments on animals sacrificed 4 to 24 hr after treatment. This length of time is probably sufficient, usually being much longer than the biological half-life *in vivo* of most of the chemicals tested. In addition, the temporal range spanned by the different experiments can be considered sufficiently narrow with regard to the scale of CBI potencies. For this reason we did not deem it necessary to modify the *in vivo*-CBI formula proposed by Lutz (24).

The *in vivo*-CBI values concerned experiments on mice and rats treated by oral or parenteral routes. Only data on liver DNA adducts were collected. Only data concerning positive results were considered. When numerous data were available for the same chemical, only the first ten values were considered.

In Vitro DNA Covalent Binding (In Vitro-CBI)

The (in vitro-CBI) was defined as:

micromole chemical bound per mole nucleotides

(millimolar chemical in incubation mixture)*(incubation time in minutes)

The in vitro-CBI values were computed from papers cited in the ICRDB Cancergrams (25); data were obtained also from Medline, Toxline, and Cancerline (National Library of Medicine, Bethesda, MD, USA: and Deutsches Institut für Medizinische Dokumentation und Information (DIMDI), Cologne, FRG). When numerous data for the same chemical were available from different papers, only the first ten were used. From a mathematical point of view, it is extremely unlikely that the average of a random sample of 10 values should be totally unrelated to the entire population of all available data. As a consequence, we considered a sample size of 10 data to be adequate. The in vitro-CBI values were computed only from experiments in which the metabolic activation was obtained with microsomes prepared from the livers of mice or rats. Different activation systems were not considered. Data for seven directly alkylating chemicals (incubations without microsomes) were also used. The considered experiments used native double-stranded calf thymus or salmon sperm DNA in the incubation mixtures. The molecular weight of the nucleotides was assumed to be equal to 309. In the same paper when more than one in vitro-CBI value was computable for the same chemical, only the highest CBI value was used. Only data concerning positive results were considered. Table 1 lists the chemicals for which we computed at least one in vitro-CBI value.

Table 1. List of the carcinogens for which an in vitro-CBI was computable.

In vitro-CBI	Liver microsomes				
	Animal species ^b	Enzymatic induction ^c	Concentrationd	Log ₁₀ (in vitro-CBI) mean value	References
Aflatoxin B ₁					
1140	R	PB	1.0	2.85	(26)
2575	R	WI	e		(27)
162	R	PB	83		(28)
490	R	PB	1.0		(29)
843	R	PB	1.0		(30)
1010	R	PB	1.1		(31)
2960	R		e		(32)
175	R		f		(33)
403	R	\mathbf{MC}	g		(34)
Aflatoxin G ₁					
41.9	R	PB	83	1.62	(28)
3-Amino-1-methyl-5	H-pyrido (4,3-b)-indo	ole; (Trp-P-2)			
7.33	R	MC	1.0	0.826	(35)
6.13	R	PCB			(36)

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Table 1. (Continued)

	Liver microsomes			• (• • • • • •	
In vitro-CBI	Animal species ^b	Enzymatic induction ^c	$Concentration^d$	Log ₁₀ (in vitro-CBI) mean value	References
Benz(a)anthracene					
0.605	R	MC	0.22	-0.218	(37)
Benzene					
1.99	M	PB	0.67	0.299	(38)
Benzo(a)pyrene	D.	140		4.0=	()
194	R	MC	0.50	1.37	(39)
20.1 3.34	R	BF	0.50		(40)
49.2	R R	MC	0.13		(41)
23.2	R	MC MC	g 0.20		(34) (42)
1710	Ř	MC	2.0		(42)
40.0	Ř	MC	h		(44)
0.258	R	MC	0.22		(37)
3.17	M	WI	0.63		(45)
62.6	R	MC	0.36		(31)
Bromoacetaldehyde					
_ 181	M		0.62	2.26	(46)
Bromobenzene	3.5				
5.36	M	PB	0.67	0.729	(47)
Bromoethanol	36		0.40	1.10	(
15.4 Carbon tetrachloride	M		0.62	1.19	(46)
0.110	M	MC	i	0.050	(10)
Chlorobenzene	IAT	MC	1	-0.959	(48)
2.84	R	PB	0.67	0.453	(49)
Cyclophosphamide	10	1.0	0.01	0.400	(43)
1.22	R	PB	0.25	0.0864	(50)
Dibenz(a,h)anthracene			0.20	0.0001	(50)
0.192	R	MC	0.22	-0.717	(37)
1,2-Dibromo-3-chloroprop	ane			···	(0.)
1.03	R		2.2	0.0128	(51)
1,2-Dibromoethane					\/
6.63	M	PB	0.67	0.587	(52)
18.7	M	PB	0.67		(53)
8.02	M		0.62		(46)
3.30	M		0.62		(54)
0.263	R		2.2		(51)
3,3'-Dichlorobenzidine	n		0.55	0.050	()
1.87 1,1-Dichloroethane	R		0.75	0.272	(55)
1,1-Dichloroethane 1.20	M	PB	0.67	0.0792	(50)
1,2-Dichloroethane	141	ГD	0.01	0.0792	(56)
8.74	R	PB	0.67	0.591	(52)
8.17	M	PB	0.67	0.031	(53)
0.827	M	• •	0.62		(54)
Diethylnitrosamine			3.3 <u>2</u>		(04)
0.552	R	PB	1.6	-0.258	(57)
Diethylstilbestrol					(/
1.33	R	MC	0.80	0.124	(58)
15,16-Dihydro-11-methylc		nanthren-17-one			
24.1	R		j	1.38	(59)
7,12-Dimethylbenz (a) anth					
20.8	R	PCB	0.25	0.711	(60)
3.41	R	PB	1.0		(61)
3.22	R	MC	0.22		(37)
6.01	R	MC	h		(44)
2.60	R	PCB	1.0		(62)
Dimethylnitrosamine	R	WI	1.6	0.792	(00)
11.8 6.39	к М	W I A	1.6 4.0	0.192	(63)
3.17	R	л	4.0 1.6		(64) (56)
Epichlorohydrin	14		1.0		(00)
0.155	R	PB	0.67	-0.810	(65)
Ethionine	10	1.0	0.01	-0.010	(00)
0.162	R	MC	i	-0.790	(66)
z-Hexachlorocyclohexane		M2.0	•	0.100	(00)
0.979	M	PB	0.20	-0.00922	(67)
y-Hexachlorocyclohexane			5. <u>_</u> 5	V. V V 400	(37)
1.49	M	PB	0.20	0.173	(67)
Hexachloroethane					(3.)
6.31	M	PB	0.67	0.800	(68)
			••••		(00)

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Table 1. (Continued)

	Liver microsomes				
In vitro-CBI	Animal species ^b	Enzymatic induction ^c	Concentration d	Log ₁₀ (in vitro-CBI) mean value	References
Isophosphamide					
21.9	R	PB	k	1.34	(69)
7-Methylbenz(c)acridine					
24.7	M	MC	1.5	1.39	(70)
3-Methylcholanthrene					
13.6	R	MC	h	1.78	(44)
266	R		2.7		(71)
Mycophenolic acid					4
1.11	\mathbf{R}		1	0.0453	(72)
N-Nitrosopiperidine					
0.169	R	PB	1.6	-0.772	(57)
Pentachloroethane					
12.9	M	PB	0.67	1.11	p
Quinoline					
4.00	R	PCB	1.0	0.602	(73)
1,1,2,2-Tetrachloroethane					
2.92	M	PB	0.67	0.465	(74)
Tetrachloroethylene	242				, , , , ,
3.74	R	PB	0.67	0.573	(75)
1.1.1-Trichloroethane	16	12	0.01	0.0.0	()
0.675	M	PB	0.67	-0.171	(76)
1,1,2-Trichloroethane	141	16	0.01	0.1.1	(,
	R	PB	0.67	1.21	(77)
16.4	r.	гь	0.01	1.21	(**)
Trichloroethylene	M	PB	0.58	-0.294	(78)
0.316	M			-0.234	(79)
0.817	M	PB	2.0		(13)
Diethylsulfate				0.400	(00)
0.347				-0.460	(80)
Dimethylsulfate				2 242	(04)
1.77				0.248	(81)
1,1'-Ethylene-bis-(1-nitro	sourea)				
2.06				0.314	(82)
Ethylmethane sulfonate					4
0.330				-0.601	(80)
0.238					(83)
0.200					(81)
Ethylnitrosourea					
0.129				-0.321	(82)
0.552					(80)
1.53					(81)
Methylmethane sulfonate	e				
2.57	-			0.420	(83)
2.69					(84)
Methylnitrosourea					• • •
4.12				0.758	(82)
2.50					(85)
18.3					(81)
Semicarbazide					` '
0.00158				-2.80	(86)

^{*}In vitro-CBI: in vitro DNA covalent binding index, as defined in "Methods."

Quantitative Correlation Studies

In a previous paper we showed that general sets of values concerning oncogenic potencies or short-term test potencies tend to display approximately a lognormal distribution (15). As a consequence, the correlation studies were made between log10 of potencies.

When more than one potency value was available for the same compound, we used the average of log₁₀ potencies. Usually, the clouds of points in graphs of \log_{10} of potency X versus \log_{10} of potency Y are also compatible with a linear regression analysis and parametric statistics (data not reported).

bM, mouse; R, rat.

^{&#}x27;A, acetone; BF, 5,6-benzoflavone; MC, 3-methylcholanthrene; PB, phenobarbital; PCB, polychlorinated biphenyls; WI, without induction.

^dThe concentration is expressed in mg protein/mL incubation mixture, unless otherwise stated. °0.50 nmole P-450/mL incubation mixture.

^f0.25 nmole P-450/mL incubation mixture.

κ' Microsomes from: 0.25; 0.7; 0.3; 0.14; 0.25 and 2.0 g of liver incubated in: 5, 6, 6, 3, 3, and 50 mL of incubation mixture, respectively, for footnotes g, h, i, j, k, l

^mIn the presence of cytosolic proteins.

[&]quot;DNA source not specified.

[°]HeLa cell DNA.

POur unpublished data.

In order to determine the effects of a possible deviation from a log-normal distribution, the different correlations were also analyzed with the nonparametric Spearman's rank correlation coefficient, as reported in Tables 2 and 3.

Results

As detailed in the "Methods" section, the main conditions adopted for normalizing the data were studies on mice and rats treated by oral or parenteral routes for carcinogenicity data, on mouse and rat liver DNA for in vivo-CBI data, and using mouse or rat livermetabolizing systems for in vitro-CBI data.

In these conditions, carcinogenicity data for 49 genotoxic chemicals (as defined in "Methods") and in vivo-CBI data for 44 genotoxic chemicals were available. They had in vitro-CBI values spanning a range more than five logs₁₀ wide, and totaled 48 compounds. However, 19 of them (i.e., chemicals No. 7-11, 17, 20, 21, 26-28, 30, 32, 34, 35, 39, 41, 43, 48) could not be used for any correlation because in vivo data were either unavailable or the chemicals were not defined as

genotoxic compounds. Thus, we were left with 29 useful compounds. In Table 1 both the *in vitro*-CBI values for each experiment and average \log_{10} (*in vitro*-CBI) values were reported; however, for correlation studies only the average \log_{10} values were used.

We wanted to consider only genotoxic compounds. The importance of this point may be illustrated by considering that during their lifetime animals subjected to chronic carcinogenicity experiments with promoters can be exposed to a background level of initiating events that are sufficient to combine with the effects of a full-promoting treatment, thus generating a detectable tumor incidence. As a consequence, even purely or prevalently promoting agents can give positive results in chronic carcinogenicity experiments in rodents. However, a study of the correlation between DNA adduct formation and carcinogenicity makes sense only if we deal with initiating agents.

In order to increase the probability that we were dealing with an initiator and not a promoter, we established the condition that a chemical could be used in our correlation study only if it was positive in at least one out of three genotoxicity tests, as judged

Table 2. Quantitative correlations among overall carcinogenicity (TD₅₀), in vivo DNA covalent binding (in vivo-CBI), and in vitro DNA covalent binding (in vitro-CBI).*

Couple of parameters	No. of chemicals	Correlation coefficient, r^{b}	Correlation coefficient per ranks, $r_s^{ extsf{d}}$
Overall correlation			
$Log (in \ vitro-CBI) = f (-Log [TD_{50}])$	26	0.44	0.30
$Log (in \ vivo-CBI) = f (-Log[TD_{50}])$	41	0.52	0.51
$Log (in \ vitro-CBI) = f (Log [in \ vivo-CBI])$	21	0.64	0.53
Correlation for the same 18 chemicals			
$Log (in \ vitro-CBI) = f (-Log [TD_{50}])$	18	0.42^{c}	0.27
$Log (in \ vivo-CBI) = f (-Log [TD_{50}])$	18	0.46^{c}	0.36
$Log (in \ vitro-CBI) = f (Log [in \ vivo-CBI])$	18	0.58	0.47

^aThe definitions of the parameters are reported in "Methods." In vitro-CBI data were obtained from the data base listed in Table 1. In vivo-CBI data were obtained from (14,24,28,33,41,50,53,65,66,68,74-76); our unpublished data were used for pentachloroethane. TD_{50} were obtained from Gold et al. and Parodi et al. (19,21).

Table 3. Quantitative correlations among liver carcinogenicity (TD₅₀), in vivo DNA covalent binding (in vivo-CBI), and in vitro DNA covalent binding (in vitro-CBI)*.

Couple of parameters	No. of chemicals	Correlation coefficient, r^{b}	Correlation coefficient per ranks, $r_{ m s}^{ m d}$
Overall correlation			
$Log (in \ vitro-CBI) = f (-Log [TD_{50}])$	13	0.64	0.55
$Log (in \ vivo-CBI) = f (-Log[TD_{50}])$	25	0.57	0.63
Correlation for the same nine chemicals			
$Log (in \ vitro-CBI) = f (-Log [TD_{50}])$	9	0.66^{c}	0.50
$Log (in \ vivo-CBI) = f (-Log [TD_{50}])$	9	0.75	0.85

^aThe definitions of the parameters are reported in "Methods." In vitro-CBI data were obtained from the data base listed in Table 1. In vivo-CBI data were obtained from (14,24,28,33,41,50,53,65,66,68,74-76); our unpublished data were used for pentachloroethane. TD_{50} were obtained from Gold et al. and Parodi et al. (19,21).

^bParametric statistical computations according to Snedecor and Cohran (87). The r values are statistically different from zero with a p < 0.05 except where otherwise specified.

 $^{^{}c}r$ values for which p < 0.10.

 $^{^{}m d}$ Nonparametric statistical computations according to Siegel (88). $r_{
m s}=$ nonparametric correlation coefficient according to Spearman.

^bParametric statistical computations according to Snedecor and Cohran (87). The r values are statistically different from zero with a p < 0.05 except where otherwise specified.

 $^{^{}c}r$ values for which p < 0.10.

 $^{^{}d}$ Nonparametric statistical computations according to Siegel (88). r_{s} = nonparametric correlation coefficient according to Spearman.

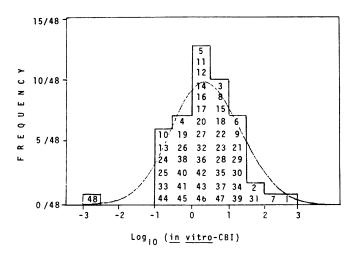


FIGURE 1. Distribution of the \log_{10} (in vitro-CBI) mean values referred to the 48 different chemicals listed in Table 1.

from the data base of Rosenkranz and Palajda (22).

We analyzed the distribution of the mean values of \log_{10} (in vitro-CBI) concerning the 48 chemicals listed in Table 1. As shown by the histogram in Figure 1, these resulting values were approximately log-normally distributed, in accordance with our previous observations (15). As a consequence of this result, we considered it acceptable to apply parametric statistics to \log_{10} of potencies.

Table 1 also details the data base of the in vitro-CBI values we computed, and gives an idea of the differences in experimental conditions present in the in vitro experiments. Although we normalized the results for drug concentration and incubation time, from the point of view of metabolic activation, the experimental conditions were definitely less homogenous than those for the *in vivo* experiments. However, as the reader will see in the following analysis, the additional statistical noise brought about by this lack of data homogeneity apparently did not play an important role. As can be seen in Table 1, for 57 out of the 87 experiments listed, it was possible to normalize the data for microsomal protein concentration. By calculating the correlation between normalized and nonnormalized data, we found r = 0.87, suggesting that the lack of this normalization is not terribly disruptive. On the other hand, by using only normalized data, we would have impoverished our data set too much.

With the data we collected, it was possible to study the quantative correlation with carcinogenicity of the *in vitro*-CBI for an overall group of 26 genotoxic chemicals, and the *in vivo*-CBI for an overall group of 41. A correlation study with more homogeneous data was possible with 18 chemicals where triplets of data on carcinogenicity, *in vivo* and *in vitro*-CBI, were available. The results are reported in Table 2.

No statistically significant differences were observed in predictivity of carcinogenicity after discarding from the correlation four directly alkylating chemicals (i.e., chemicals No. 44-47 of Table 1), whose data were available both for *in vitro-CBI* and *in vivo-CBI*.

As previously reported, quantitative correlation levels with carcinogenicity around 0.4-0.5 are very common in most of the short-term tests (15). In this respect the results reported in Table 2 seem to suggest that DNA adduct formation can contribute to our knowledge of genotoxicity and the initiating potential of a given chemical with a degree of efficacy similar to that of the most common short-term tests. This conviction is reinforced by the fact that the differences in quantitative predictivity previously found for different short-term tests never reach a level of clear statistical significance (15).

In a recent paper, Lutz reported a correlation level with carcinogenicity of the *in vivo*-CBI with r=0.81 (14). Compared with our r=0.52, this difference is statistically significant (p=0.03). However, it has to be remembered that we considered 12 more chemicals in addition to the 29 considered by Lutz, and we also considered additional experiments for those same 29 chemicals. Moreover, Lutz considered only carcinogenicity results obtained in the same species used for DNA adducts.

The results obtained with the Spearman's test were not very different from those obtained with the parametric approach. The small difference observed is probably related more to the small size of the set considered than to an important systematic deviation from normality of the log of potencies of the general population of data.

As reported in Table 2, we have also investigated the quantitative correlation existing between in vitro-CBI and in vivo-CBI data. The correlation coefficient with r value around 0.6 found in this study appears relatively high if compared with r values around 0.4, previously found when comparing different short-term tests with one another (9,10). This finding confirms that the highest correlation is reached with tests that use the same biological endpoint. As an example, in a previous study, the in vivo alkaline DNA elution test, which looks at the endpoint DNA damage, appeared to be correlated with the *in vivo*-CBI by r = 0.66. By contrast, it was correlated by lower r values around 0.3, with tests that use different endpoints such as Ames' test and in vitro cell transformation test (9).

In Table 3 we reported the correlation between DNA adducts and carcinogenicity in the liver. In this case the sample size becomes further reduced. The differences between r values obtained for overall tumors and the r values obtained for liver tumors are not statistically significant. However, a general trend seems to emerge suggesting that the identity of the target organ perhaps plays some role in determining

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the correlation level. Even in this case, the predictivities of *in vitro*-CBI and *in vivo*-CBI are very similar.

Discussion

In vivo DNA binding has already been proposed by Lutz (14, 24) as a valid short-term test for assessing genotoxicity.

To our knowledge, this is the first time that in vivo and in vitro covalent bindings have been compared in terms of their predictivity of carcinogenicity. In vitro tests are usually less expensive and more rapid than in vivo tests. In addition, the use of in vitro tests can significantly reduce the number of animals used to assess the initiating potential of a given chemical as a carcinogen. On the other hand, the general knowledge that we have about absorption, metabolism, and catabolism of chemicals leads us to suspect that in vivo assays could be more predictive than assays in vitro. As a matter of fact, the papers mentioned in the "Introduction" (16-18) seem to suggest that for DMN and 1,2-DMH, two important classical initiating carcinogens, the in vivo assay is indeed much better than the in vitro assay.

However, this difference is no longer evident when we look at the globality of the 30 to 40 chemicals that we have examined in this work. Admittedly, the sensitivity of the experimental approach adopted for comparing in vitro and in vivo predictivity is not very great (a larger data base was not available); hence, it remains possible that the in vivo assay is slightly more predictive than the in vitro assay, as suggested by the small differences in correlation coefficents found in our results. We have to stress that only limited conclusions can be justified by a data base of the size mentioned above.

In a recent study Tennant et al. (5) found the qualitative correlation between the carcinogenicity in mice and rats to be only 67%. Given that even the level of this type of internal correlation is limited, we cannot expect especially high correlation levels between short-term tests and carcinogenicity. In this perspective the levels of predictivity found in our study can be considered to be reasonably good.

If we compare the *in vitro*-CBI evaluation (in the presence of liver microsome activation) with other tests, the quantitative predictivity of this assay seems to be similar to that of other classical short-term tests, such as Ames' test, SCEs in mouse bone marrow, morphological transformation *in vitro*, DNA fragmentation *in vivo* (liver), and DNA repair *in vitro* (hepatocytes) (15).

In conclusion, in vitro-CBI seems to have a predictivity similar to most short-term tests and is not clearly inferior to in vivo-CBI. We think that in vitro-CBI can be proposed as another short-term test in its own right, especially in the perspective of reducing the number of animals used in toxicity studies.

Our data could also suggest that for other types of short-term tests (mutagenicity and chromosomal damage), in vivo versions of the tests are not necessarily dramatically more predictive than the corresponding version in vitro (with liver microsome activation). This could be a subject which deserves further investigation.

Finally, it could be interesting to compare in vitro and in vivo-CBI for a homogeneous class of chemicals. In this respect we have started investigation with the family of chloroethanes.

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